

Allen & Walker (1987)

Limited penetration depth in soil system with Freundlich sorption and first-order degradation

Jos Boesten



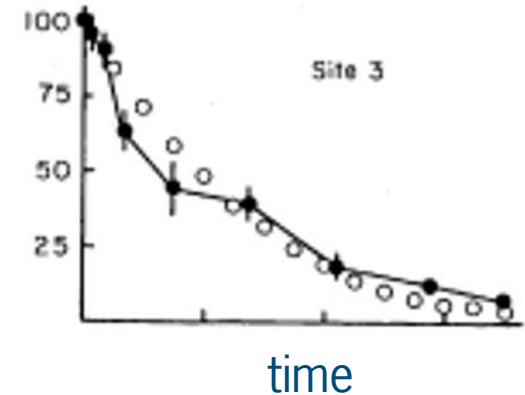
Allen & Walker (1987)

- 18 UK agricultural soils from top 10 cm
 - 14 clay loams or clays
 - limited range of textures considering other EU countries
 - organic carbon range: 0.7-2.4%
- laboratory studies with three herbicides on
 - *DegT50* at 20°C and pF = 2.5
 - Freundlich isotherm: K_f and N
- first such data set including N
- their aim: predict *DegT50* from soil properties
 - more than decade before spatially distributed modelling became popular
 - so far ahead of their time
 - prediction appeared to be difficult

Allen & Walker (1987)

Why favorite data set over 30 years ?

- Ph D supervisor Allan Walker guarantee for high-quality degradation rate measurements
 - some **75** field tests of Walker persistence models between 1973 and 1987
 - long, long before validation terminology and good-modelling practice guidance was developed
- literature information on variation in substance properties: variation also due to different methodologies
 - which part real and which part caused by scientists ?
 - Wauchope 2002 review sorption: $\max K_{oc} / \min K_{oc} = 100$?
 - this paper: 'true variability'



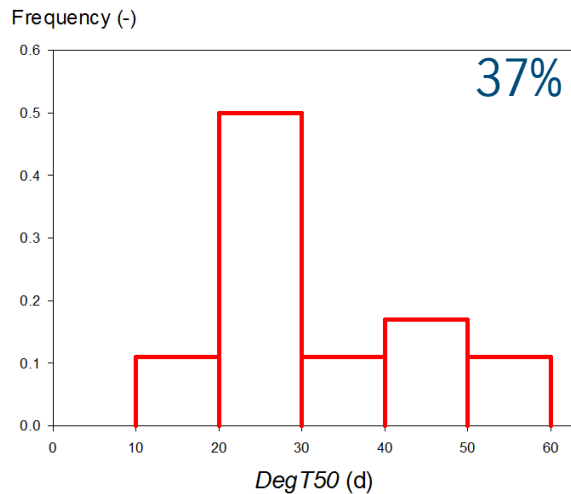
Allen & Walker (1987)

My use of the data set:

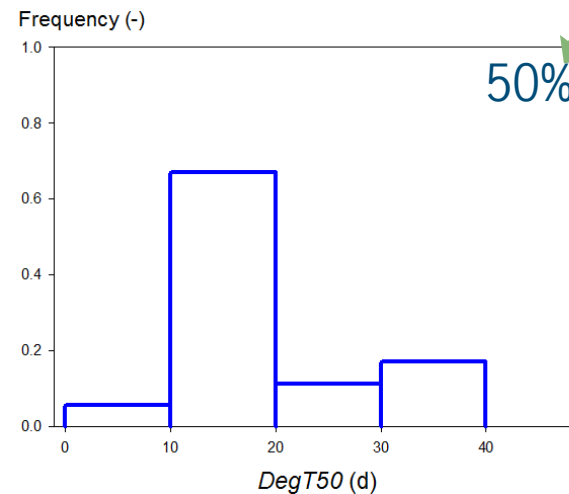
- underpinning CVs of K_{oc} and $DegT50$ for spatially-distributed exposure assessments at EU level
 - observed variability minimum level for zonal or EU level
- explore causes of variation in K_{oc} and $DegT50$
- Freundlich exponent N (pesticide property in EU guidance)
 - variability between soils
 - testing to which extent N is soil or pesticide property

Allen & Walker (1987)

metamitron



metazachlor

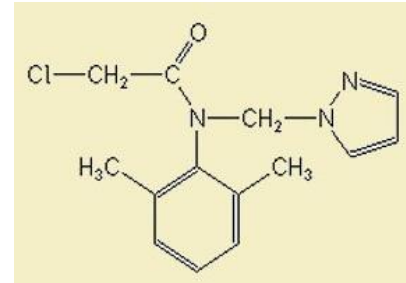
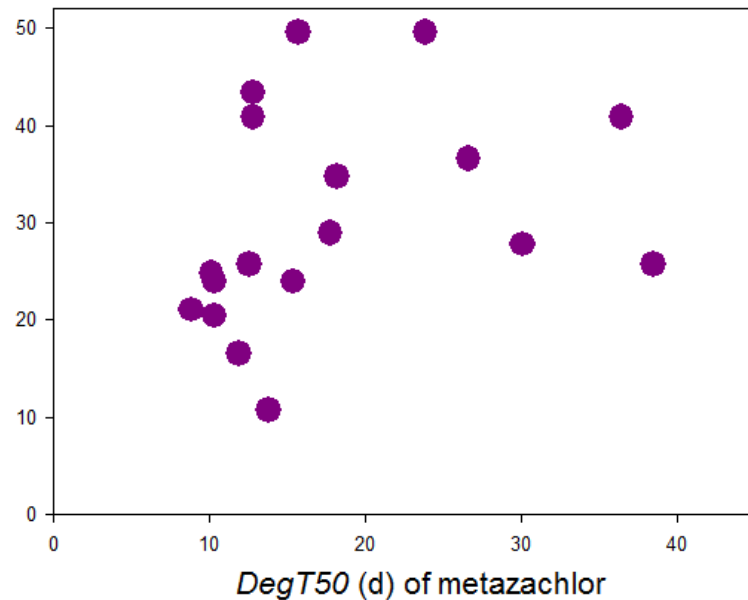


Variability in degradation half-lives

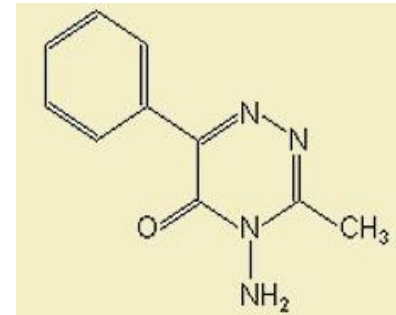
EFSA guidance assumes lognormal distribution with CV of about 50%

Allen & Walker (1987)

DegT50 (d) of met amitron



metazachlor



met amitron

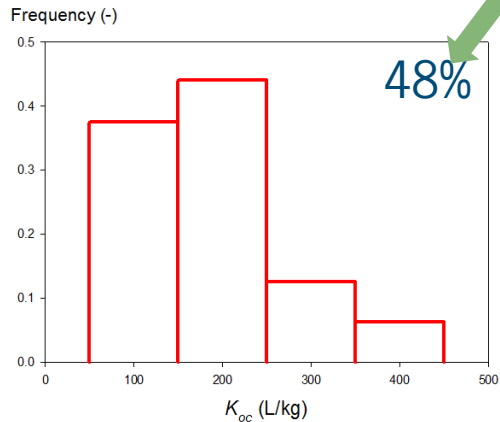
correlation coefficient 0.30 : not significant

so spatially-distributed predictions of *DegT50* difficult

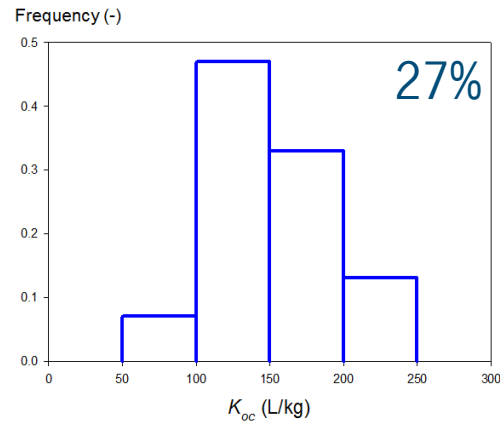
Allen & Walker (1987)

CV

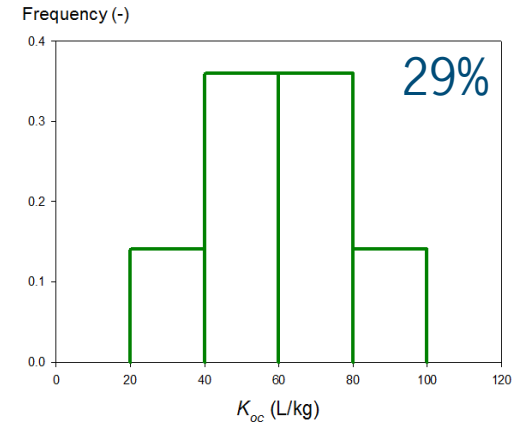
metamitron



metazachlor



metribuzin

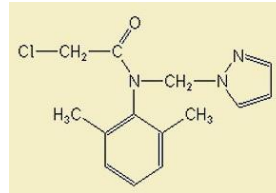


Variability of K_{oc}

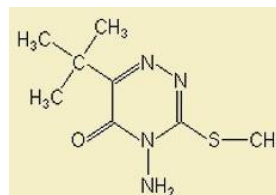
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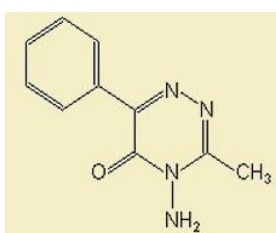
metazachlor



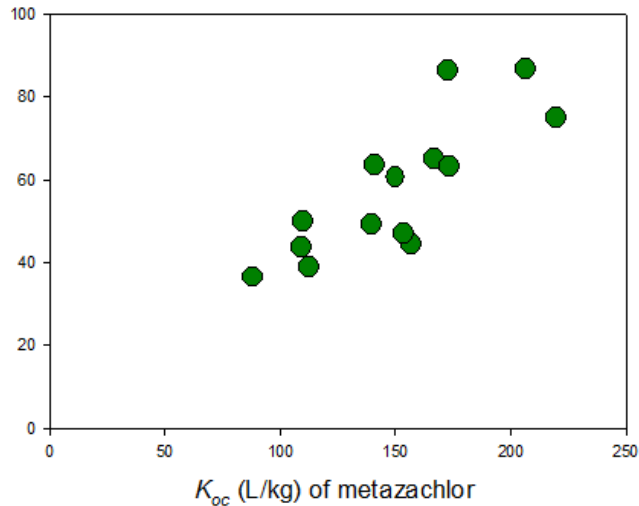
metribuzin



metamitron

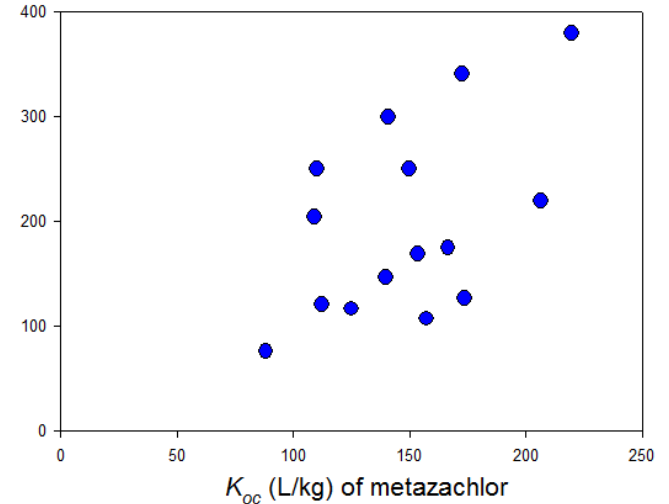


K_{oc} (L/kg) of metribuzin



strong correlation (0.82)

K_{oc} (L/kg) of metamitron

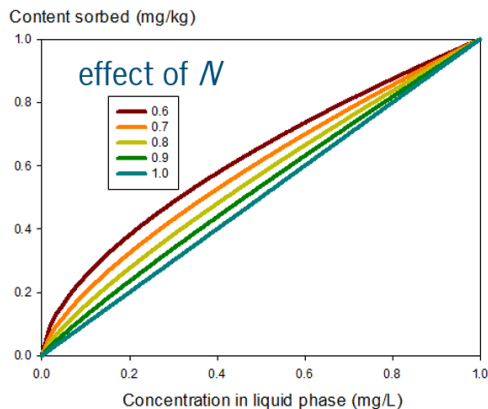
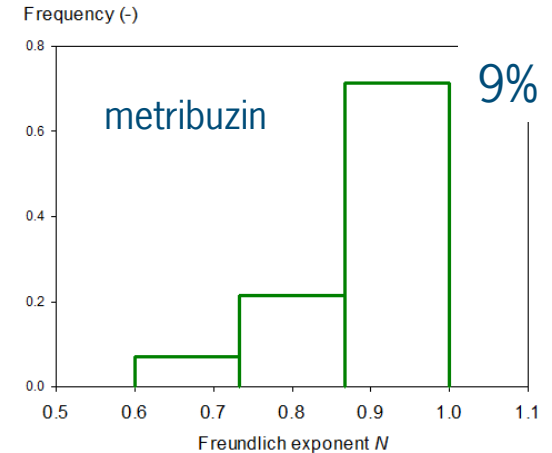
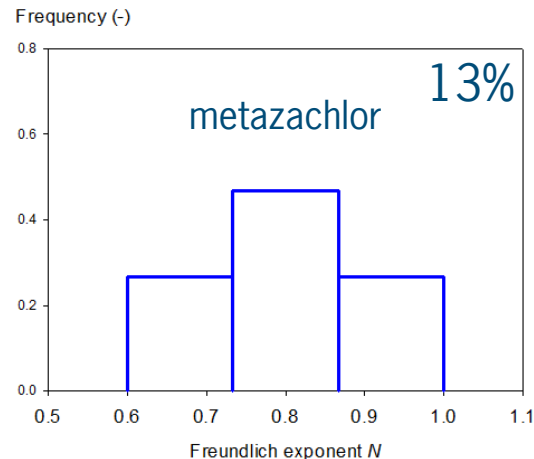
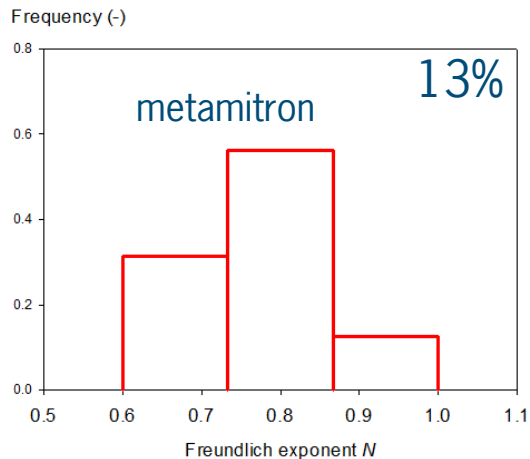


weak correlation (0.52)

K_{oc} values correlated ?

(then probably similar sorption sites)

Allen & Walker (1987)



Variability in Freundlich exponent

wide range for all three

metribuzin lower than other two

CV of three pesticides and single soil on average 12%

further statistical analysis needed to test hypothesis whether N is soil or pesticide property

Allen & Walker (1987)

- dataset very valuable for scientists interested in
 - pesticide degradation and sorption processes
 - spatially-distributed modelling of pesticide exposure



- my talk: effect of interaction between non-linear sorption and degradation on leaching behaviour
 - Allen & Walker one of most valuable sources for magnitude of non-linearity (N)

Outline

Limited penetration depth in soil system with Freundlich sorption and first-order degradation

- Introduction
- Behaviour of closed Freundlich-SFO system
- Leaching in uniform soil system
- Evidence for Freundlich-SFO behaviour
- Conclusions

Introduction

- models for FOCUS groundwater scenarios all based on Freundlich isotherm and first-order degradation of total pesticide concentration in soil system, further called Freundlich-SFO system (Single First Order)
 - PELMO
 - PEARL
 - PRZM
 - MACRO
- Freundlich-SFO approach is a cornerstone of EU leaching assessment since 2000

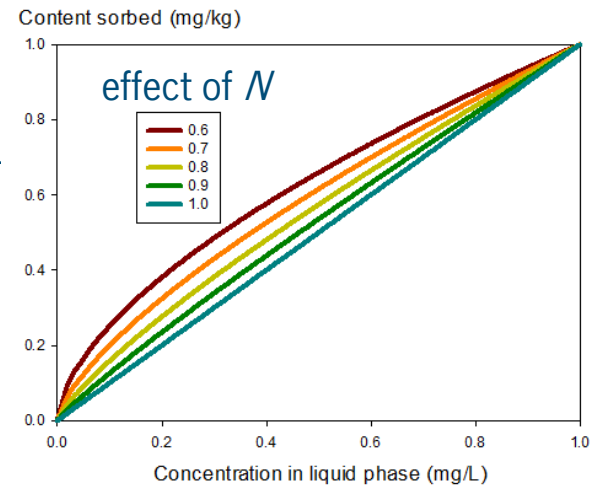
Introduction

Freundlich-SFO system

■ Freundlich sorption isotherm

$$X = K_F c_{ref} \left(\frac{c}{c_{ref}} \right)^N$$

X content sorbed (mg/kg)
 K_F Freundlich sorption coefficient (L/kg)
 N Freundlich exponent (-)
 c concentration in liquid phase (mg/L)
 c_{ref} reference value of c (mg/L)



■ SFO = single first-order kinetics

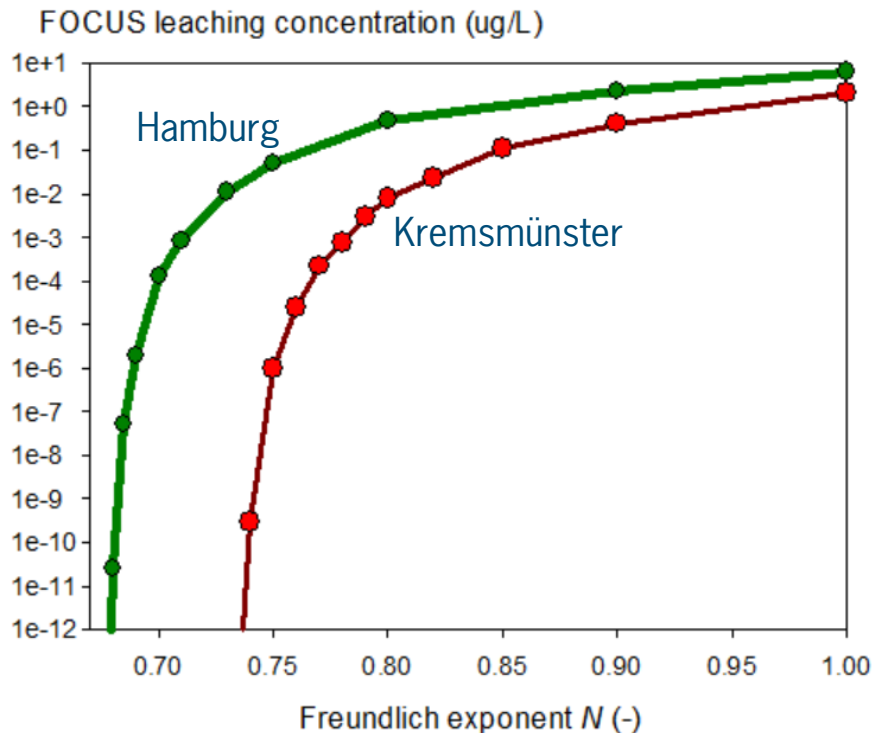
$$c^* = \theta c + \rho X$$

c^* concentration in soil system (mg/L)
 θ volume fraction of water (-)
 ρ dry bulk density (kg/L)

$$\frac{d c^*}{d t} = -k c^*$$

t time (d)
 k first-order rate coefficient (1/d)

Introduction



- FOCUS leaching concentration at 1 m depth
 - simulated with PEARL
- two pesticides applied in winter cereals in Hamburg and Kremsmünster
 - any pesticide-scenario combination shows same shape
- sensitivity to N : at some point sharp decline to zero
- why ?

Behaviour of closed Freundlich-SFO system

- e.g. incubation system for measuring degradation rate in top soil
- what is time course of fraction in liquid phase ?
- analytical approximation (ignoring mass in liquid phase):

$$f_{liq} = f_{liq,0} \exp\left(-k t \left(\frac{1-N}{N}\right)\right)$$

- f_{liq} fraction in liquid phase (-)
- $f_{liq,0}$ f_{liq} at $t=0$ depending on sorption coefficient etc. (-)
- k degradation rate coefficient (d^{-1})
- t time (d)
- N Freundlich exponent (-)

- calculations for system with $DegT50 = 200$ d and $K_F = 3$ L/kg

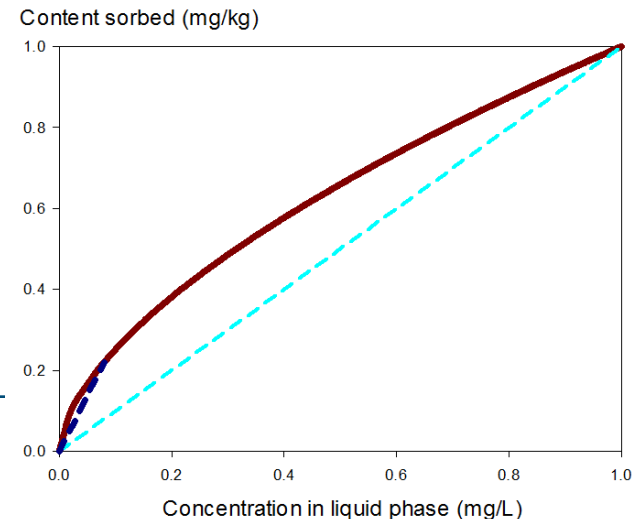
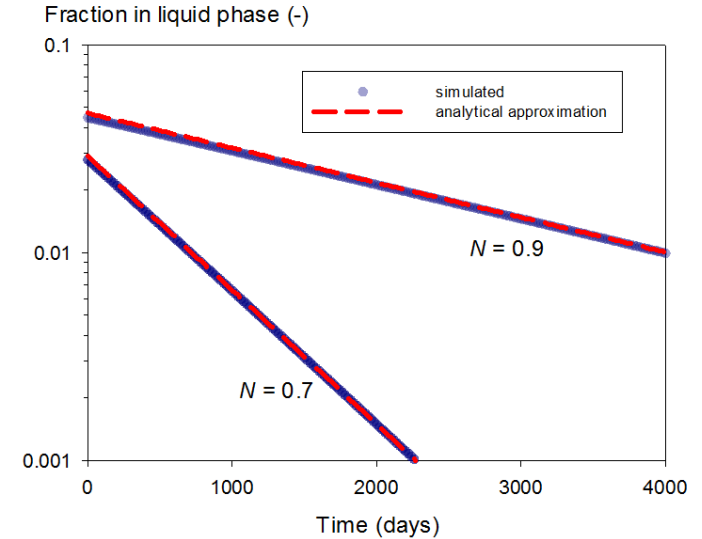
Behaviour of closed Freundlich-SFO system

- $DegT50 = 200$ d and $K_F = 3$ L/kg
 - analytical approximation works well

$$f_{liq} = f_{liq,0} \exp\left(-k t \left(\frac{1-N}{N}\right)\right)$$

N	$(1-N)/N$
0.7	0.43
0.8	0.25
0.9	0.11
1.0	0

- decrease faster for higher N
 - $N = 1$ then f_{liq} constant
- background of decrease of f_{liq} :
concentration decrease in Freundlich system leads to shift to solid phase
- background of exponential decrease of f_{liq} :
content sorbed decreases exponentially



Outline

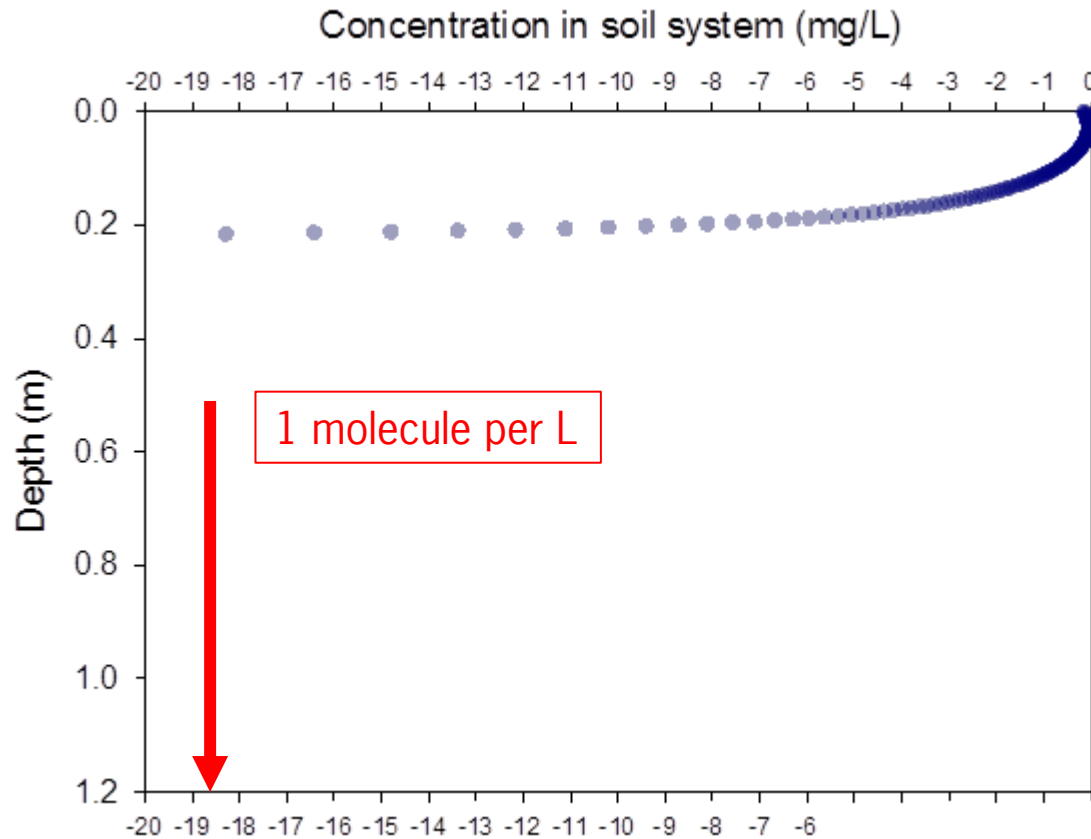
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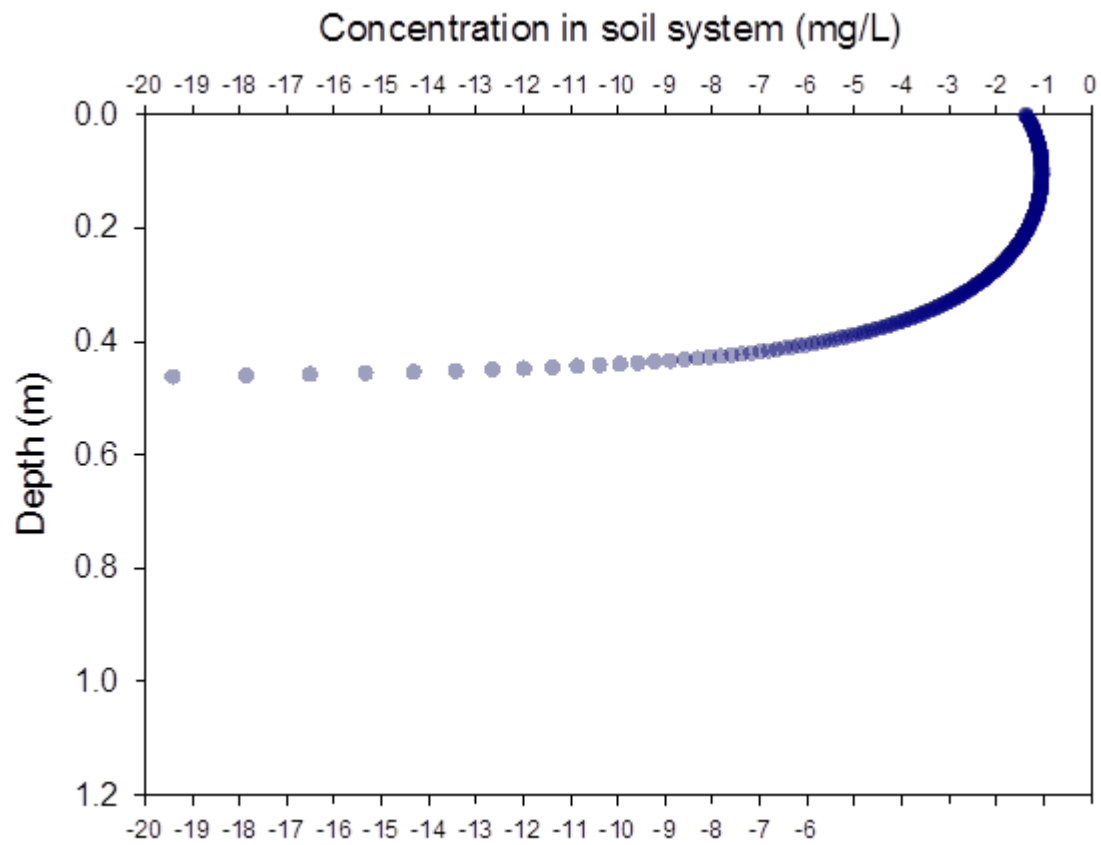
Leaching in uniform soil profile

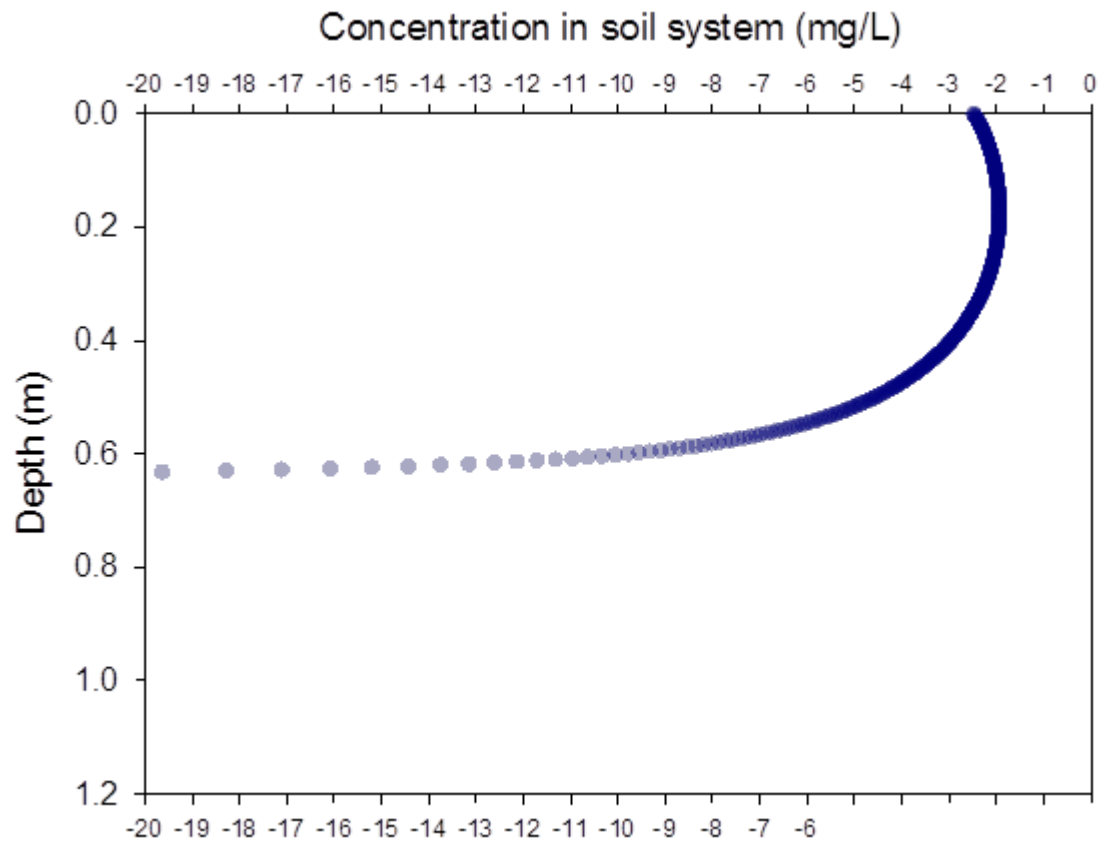
- simplified version of PEARL
- soil profile with uniform properties
 - volume fraction of water 0.25
 - water flow rate 1 mm/d
 - dry bulk density of 1.5 kg/L
 - dispersion length of 5 cm
 - no plant uptake
 - degradation and sorption uniform with depth
 - numerical compartments of 2 mm
 - single pesticide application of 1 kg/ha
- results shown for following pesticide properties
 - degradation rate based on $DegT50 = 200$ d
 - Freundlich isotherm parameters: $K_F = 3$ L/kg and $N = 0.9$
- calculations for range of other pesticide properties show qualitatively always same result

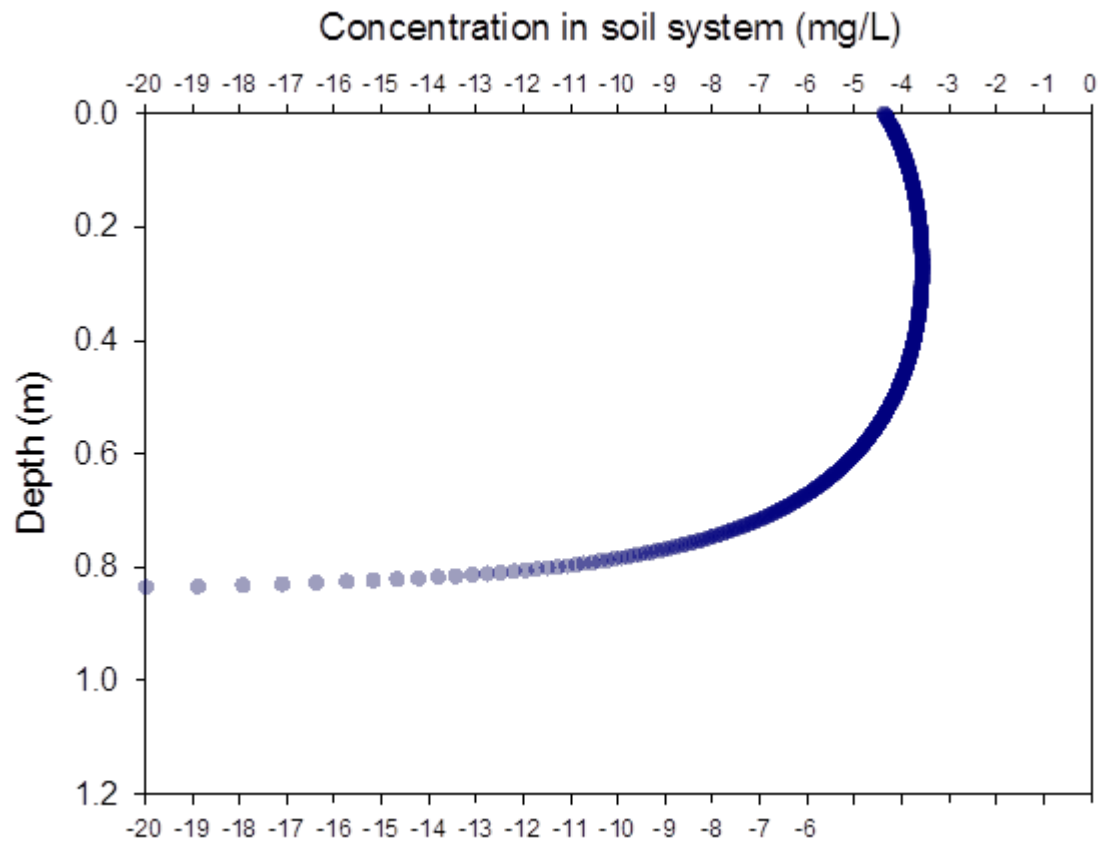
11 profiles of concentration in soil system for times increasing from 100 to 9000 d

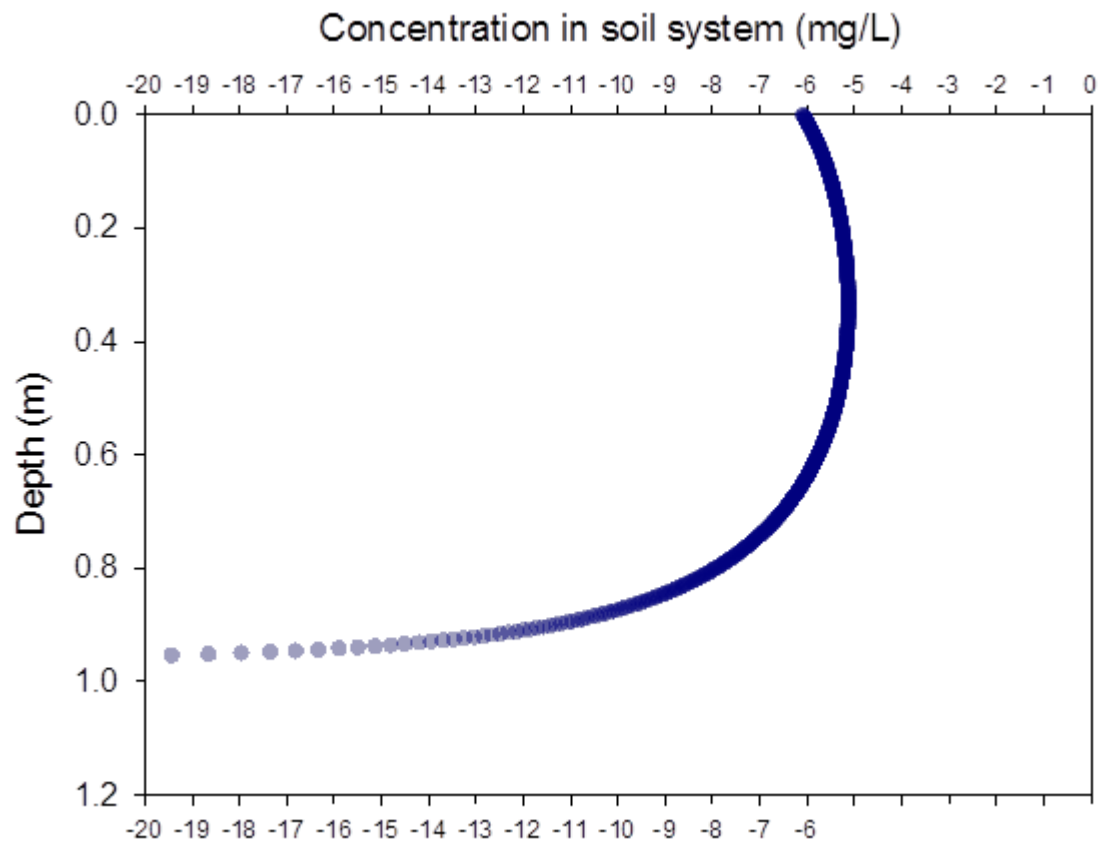


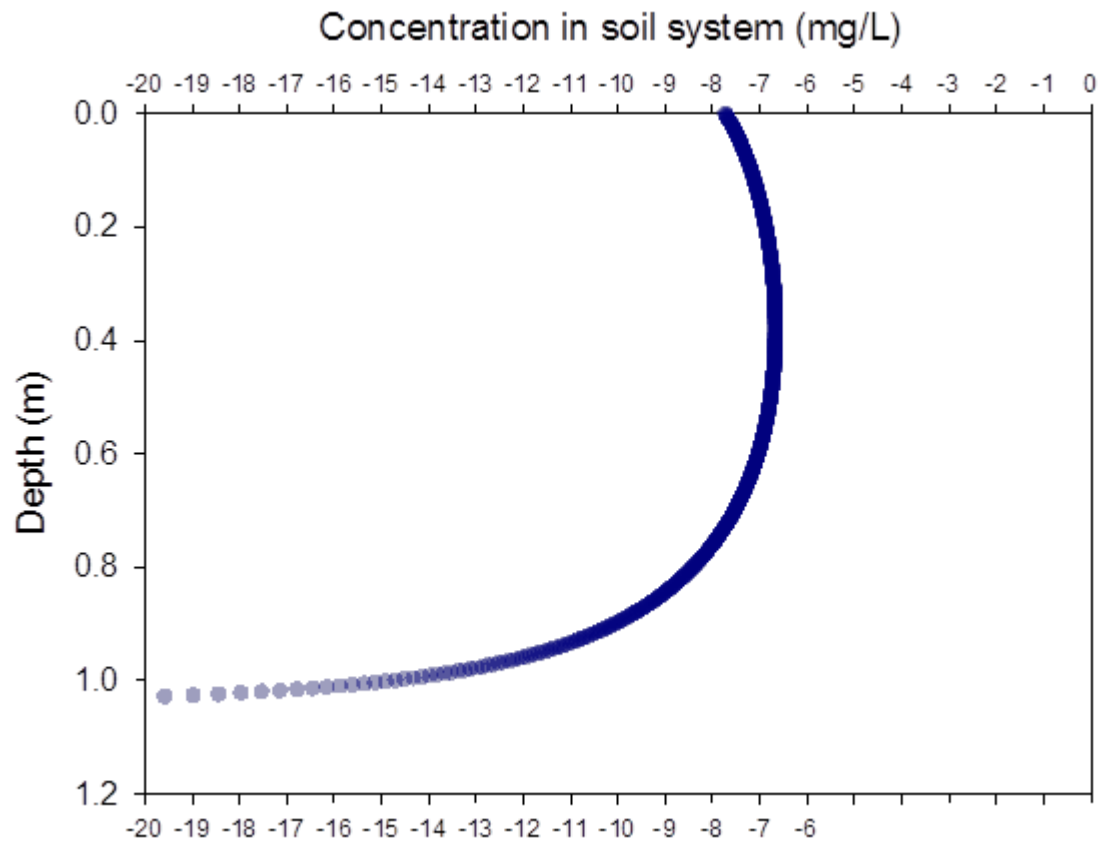
concentration axis from 10^0 to 10^{-20} mg/L
to demonstrate principle

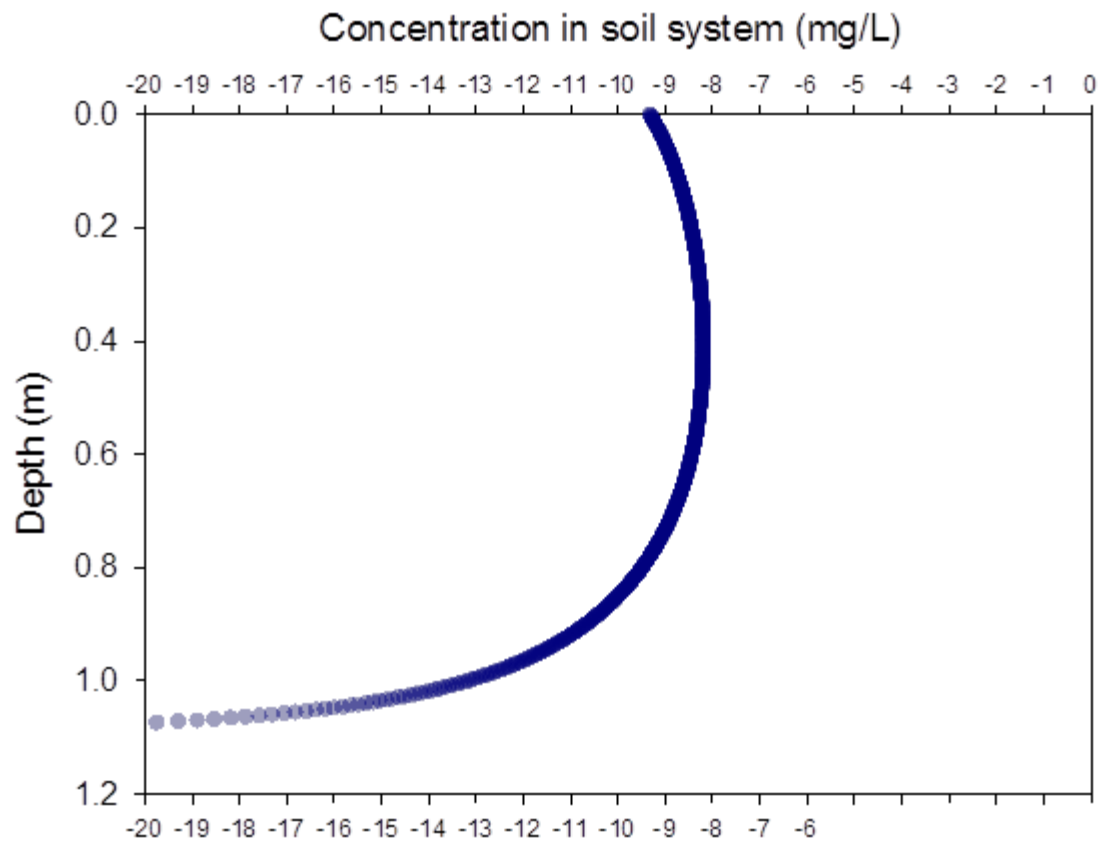


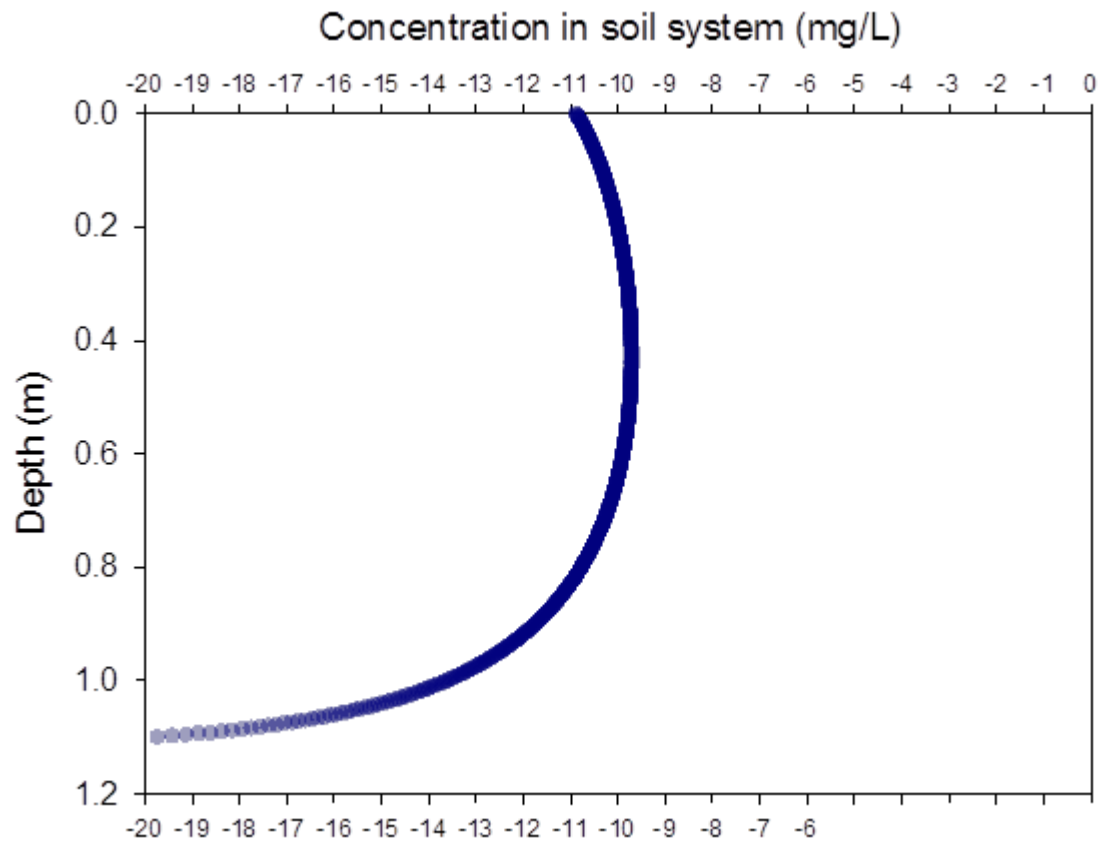


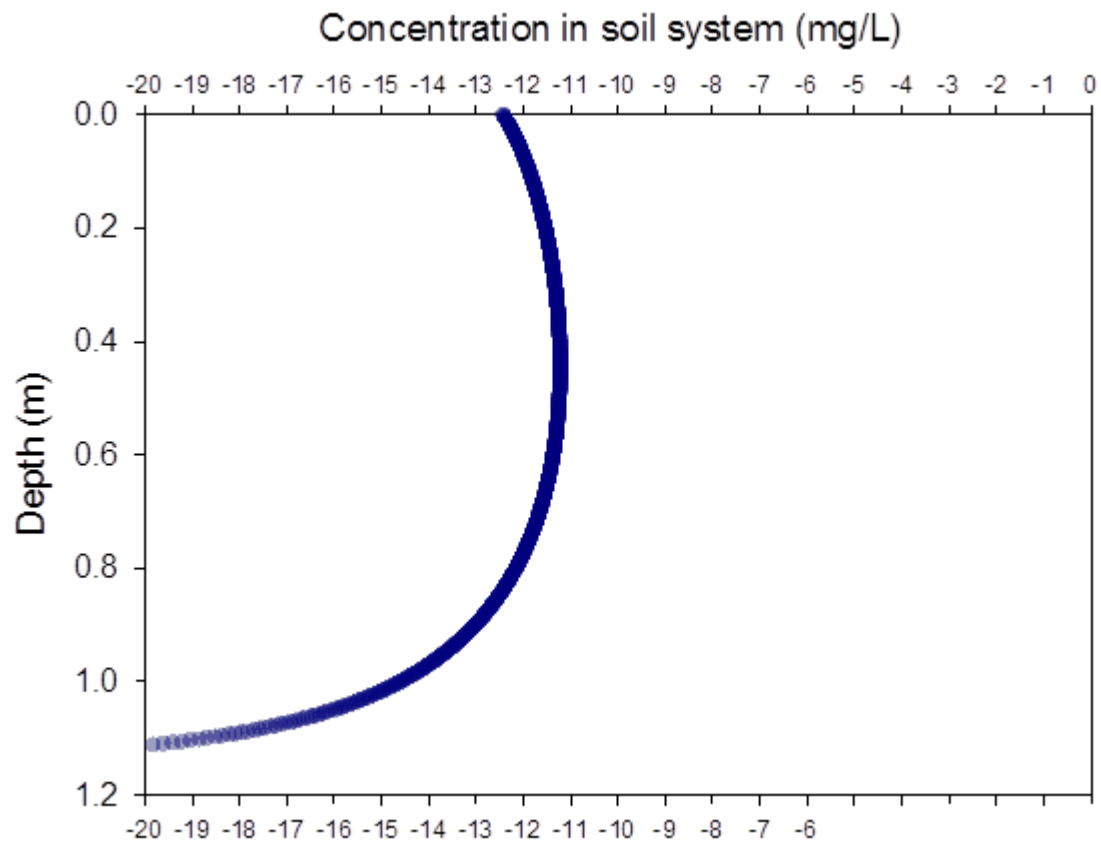


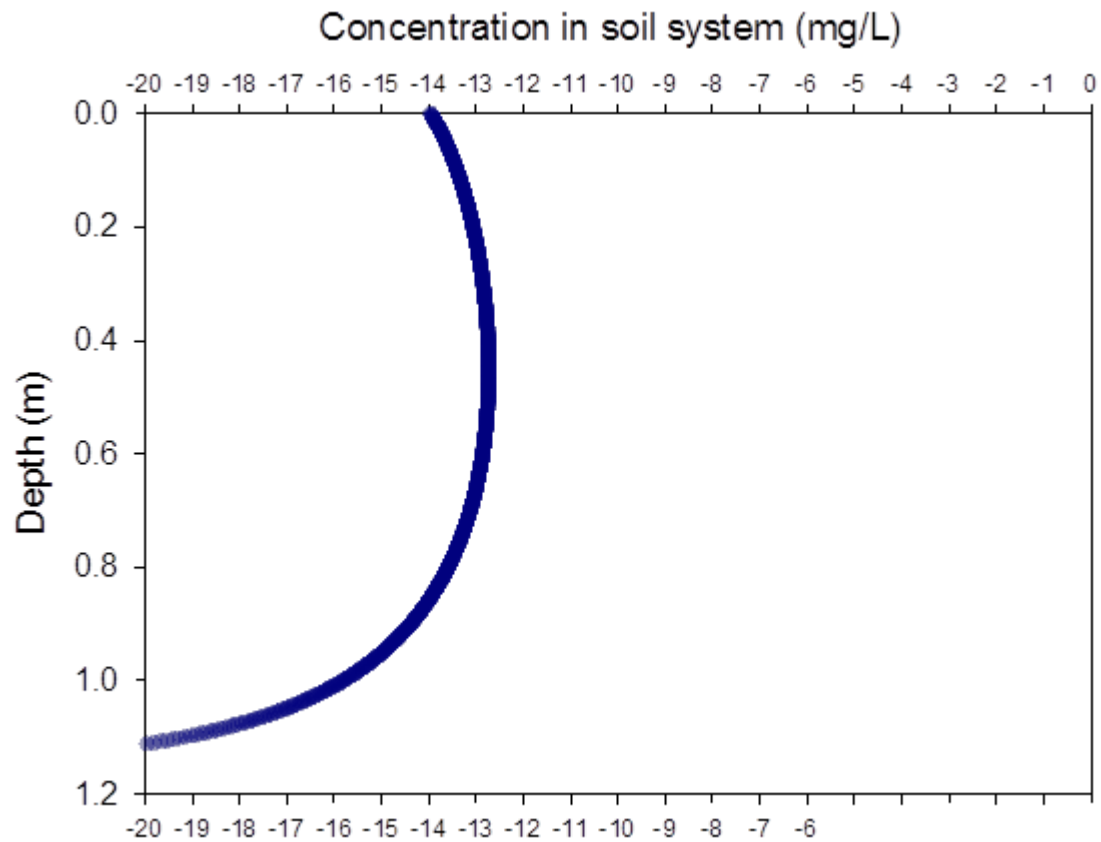


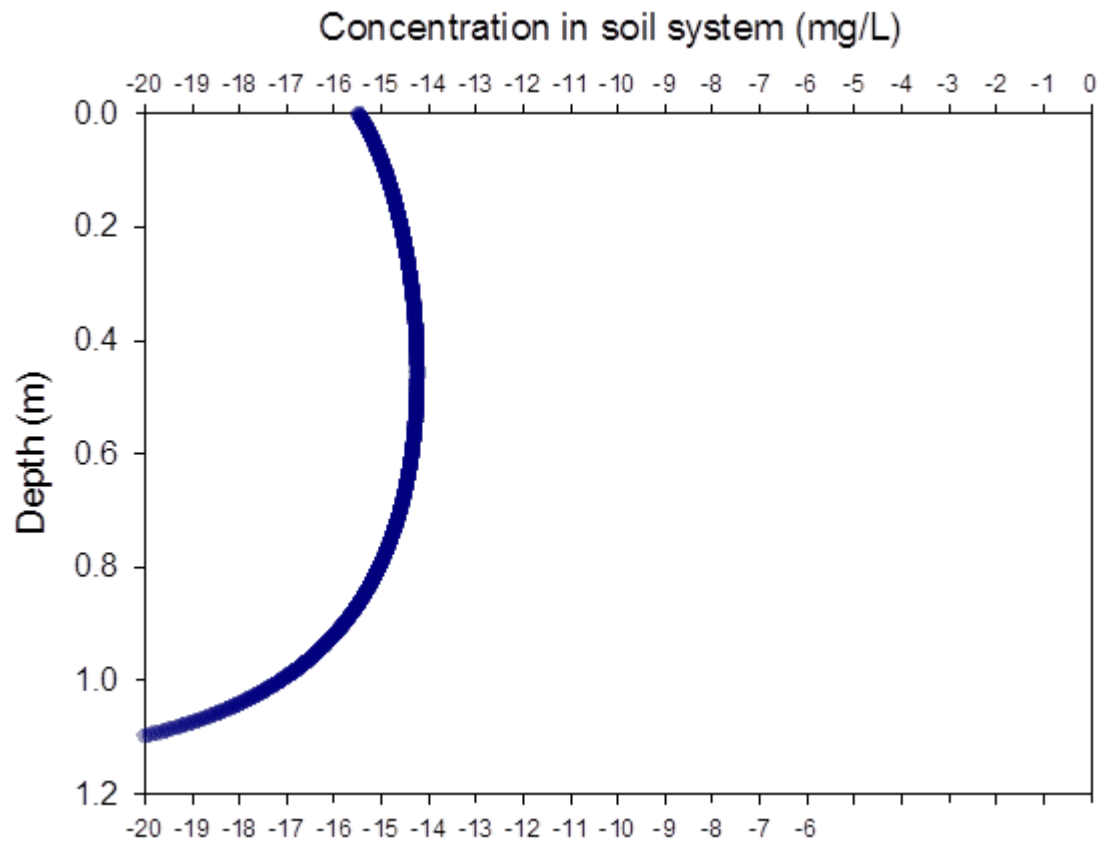


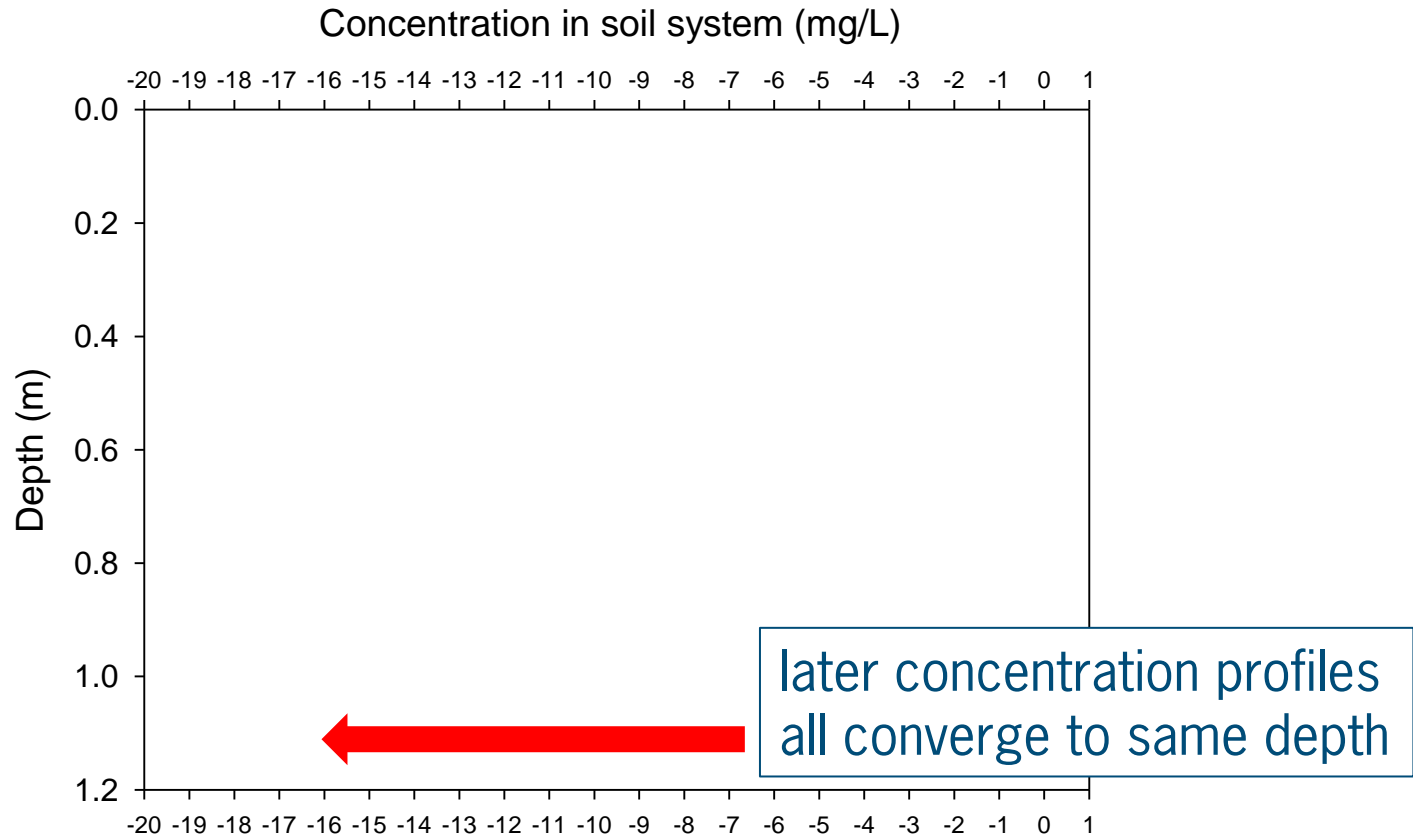




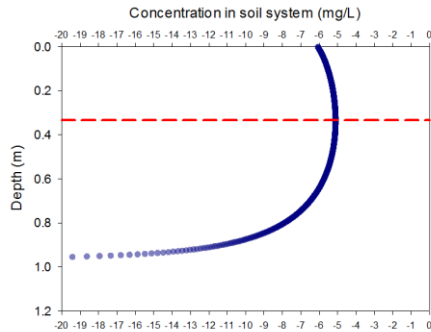






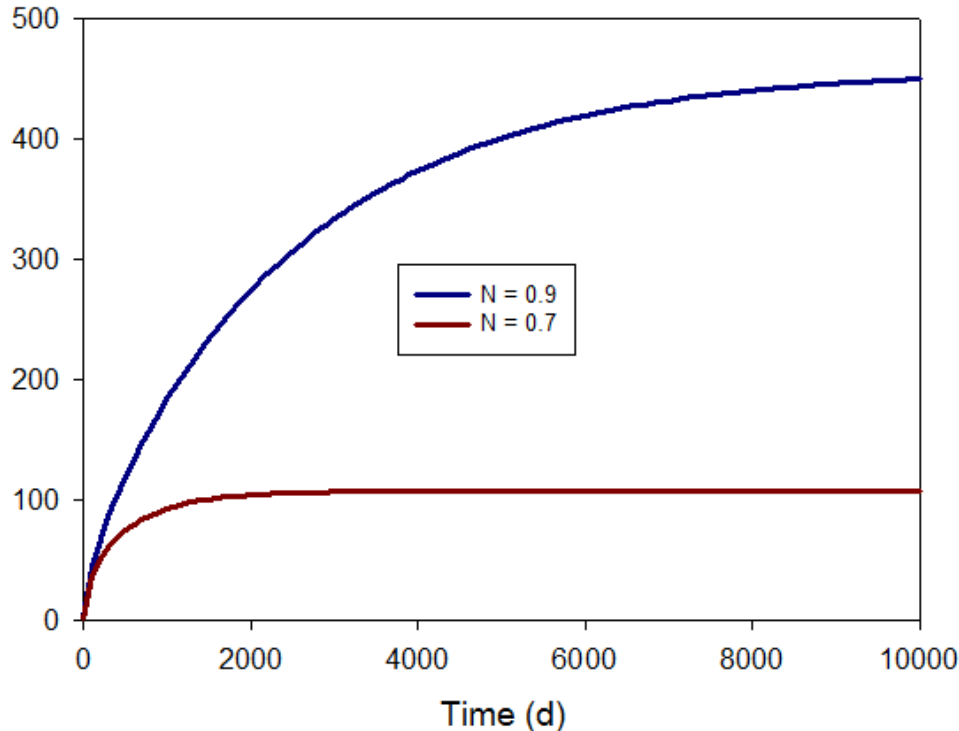


Leaching in uniform soil profile



average penetration depth:
50% above and 50% below this depth

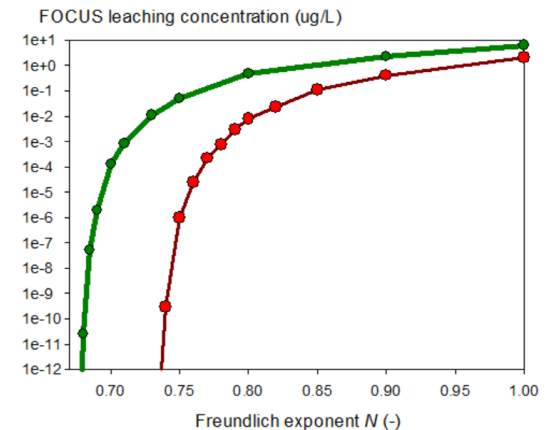
Average penetration depth (mm)



- approaches plateau
 - consistent with behaviour of concentration profiles
- for $N = 0.7$ plateau is shallower and reached quicker

Leaching in uniform soil profile

- so pulse in Freundlich-SFO leaching system with constant properties has finite penetration depth
 - beyond which no pesticide molecule will ever pass
- this causes probably the drastic drop in FOCUS leaching concentration when N decreases

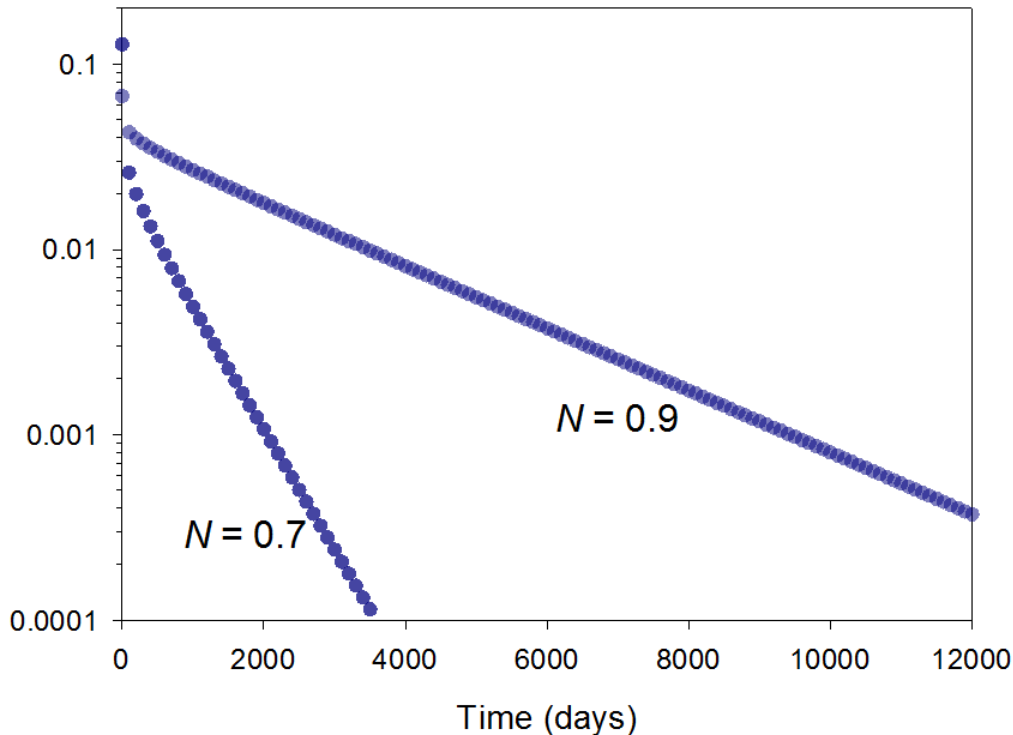


Leaching in uniform soil profile

- finite penetration depth, so speed of pulse goes to zero
- this speed is proportional to fraction of pesticide in liquid phase
 - only molecules in liquid phase move
- so let us examine fraction of pesticide in liquid phase for total mass of pesticide present in this leaching system

Leaching in uniform soil profile

Fraction in liquid phase in soil profile (-)

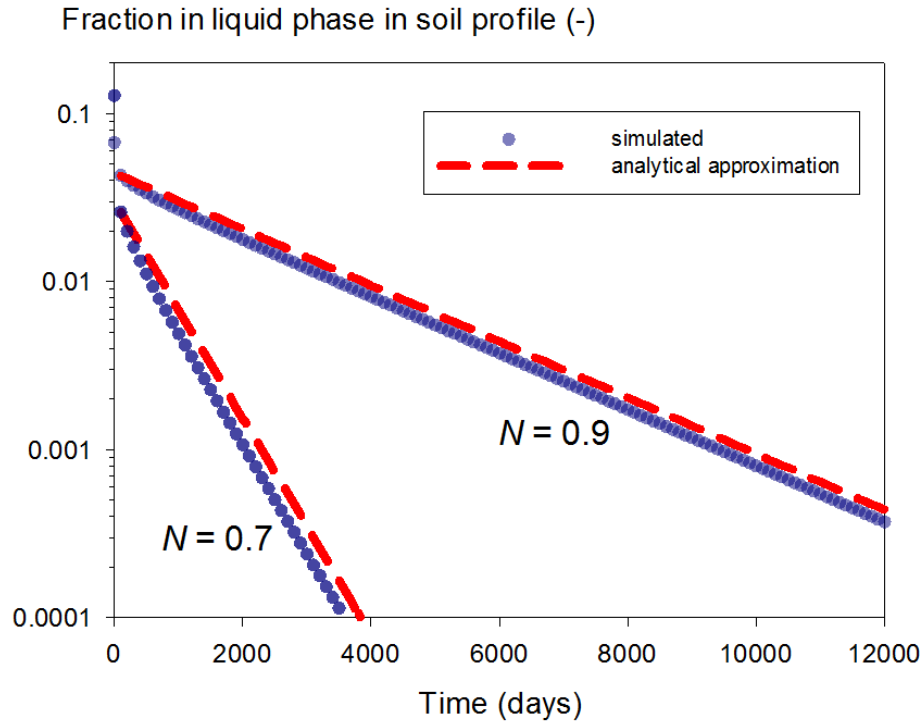


- exponential decrease similar to closed system

$$f_{liq} = f_{liq,0} \exp\left(-k t \left(\frac{1-N}{N}\right)\right)$$

- k and N are known but $f_{liq,0}$ unknown so $f_{liq,0}$ estimated from start of linear phase

Leaching in uniform soil profile



$$f_{liq} = f_{liq,0} \exp\left(-k t \left(\frac{1-N}{N}\right)\right)$$

- fraction in liquid phase in soil profile decreases at about same speed as in closed system
 - after an initial phase with a more rapid decline
- exponentially decreasing fraction in liquid phase gives exponentially decreasing speed of the pulse
- therefore a pulse has a finite penetration depth in a Freundlich-SFO soil system

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Evidence for validity of Freundlich-SFO concept

- finite penetration depth after infinite time counterintuitive
- sound underpinning needed of this cornerstone of EU leaching assessment
- Freundlich-SFO has historical roots in pesticide world
 - hundreds of measurements of degradation rate measured by extraction with solvent from 100% to some 5% show nice SFO behaviour
- problem: for simulating leaching at 0.1-ppb level the models have to extrapolate orders of magnitude below this 5%

Evidence for validity of Freundlich-SFO concept

- H_0 : hypothesis Freundlich-SFO: degradation rate proportional to concentration in total soil
 - for most substances equivalent to: rate proportional to amount sorbed

- H_a : alternative hypothesis: degradation rate proportional to concentration in liquid phase
 - my perception: main-stream microbiological thinking since about 1985
 - considerable amount of indirect evidence (e.g. paraquat behaviour)
 - very large consequences for leaching assessment
 - Beltman et al. (2008) Water Resour. Res. 44, W05417

Evidence for validity of Freundlich-SFO concept

- degradation rate measurements needed that can distinguish between H_0 and H_a
 - not easy, but doable: e.g. study substances with low N values that show very rapid microbial degradation
- to best of my knowledge, no measurements are available that support preference of H_0 over H_a
 - but I am happy to be proven wrong

Conclusions

- leaching of pesticide pulse in uniform Freundlich-SFO soil system results in limited penetration depth
 - which decreases with increasing curvature of isotherm
- this is caused by exponentially decreasing fraction in liquid phase

Conclusions

- this exponentially decreasing fraction is caused by Freundlich-SFO assumption that degradation rate is proportional to total amount in soil instead of amount in liquid phase
- experimental tests of this Freundlich-SFO degradation rate concept required because
 - above results are counterintuitive
 - indirect evidence in favour of alternative concept

Thank you for your attention !

